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## OXYGEN RELATED TRANSPORT PHENOMENA NEAR THE POLYSILICON/SIO<sub>2</sub> INTERFACE ON SILICON DURING PHOSPHORUS DIFFUSION

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The structural modifications near the polysilicon/SiO<sub>2</sub> interface on Cz silicon substrate during the phosphorus diffusion from a liquid POCl<sub>3</sub> source were investigated by cross-silicon transmission electron microscopy (XTEM). A thick (1.5  $\mu$ ) polysilicon layer was deposited by the LPCVD method at 620°C and a phosphorus prediffusion at 1030°C for 15 min and a diffusion at the same temperature for 20 min were subsequently performed. The SiO<sub>2</sub> precipitation process from the dioxide layer of 16 nm thickness was discussed taking into account both the oxygen atom transport from the SiO<sub>2</sub> interfacial layer and from silicon bulk of the substrate, in interaction with the silicon self-interstitial atoms generated by the phosphorus diffusion in silicon and by the polysilicon oxidation during the phosphorus prediffusion.

#### 1. INTRODUCTION

Recently it was described the influence of the phosphorus doping on the restructuring process of the polysilicon layers on silicon during phosphorus diffusion in the polysilicon/silicon system from a POCl<sub>3</sub> liquid source [1]. Such a system is useful because it improves the electrical parameters of the npn transistors and their reliability, as well as for microsystem applications [2]. It was shown [1] that the phosphorus diffusion influences the restructuring process of the polysilicon layers, determining the enhancement of the grain growth due to the silicon self-interstitial injection induced by phosphorus diffusion in the silicon grain bulk and in the silicon substrate. During the phosphorus diffusion from the liquid POCl<sub>3</sub> source, self-interstitial atoms are generated by the oxidation

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process associated to the phosphosilicate (PSG) growth on the top of the polysilicon layer. Restructuring process also influences the doping properties, especially near the polysilicon/SiO<sub>2</sub> interface on silicon [2].

In this paper we discuss new results concerning the behaviour of the polysilicon/SiO<sub>2</sub>/silicon system during the phosphorus diffusion from a POCl<sub>3</sub> source (prediffusion step) and during the subsequent diffusion treatments, especially consisting in the degradation of the thick (16nm) interfacial layer and SiO<sub>2</sub> precipitation in the silicon substrate near the polysilicon/SiO<sub>2</sub> interface. A mechanism of this precipitation is proposed. Such a mechanism involves the oxygen transport from the initial continuous SiO<sub>2</sub> film to the silicon region near the polysilicon/SiO<sub>2</sub> interface and determines the extension of the SiO<sub>2</sub> precipitated film to the silicon bulk. The influence of the oxygen atoms from Cz type silicon wafers on the SiO<sub>2</sub> precipitation near the polysilicon/SiO<sub>2</sub> interface is also discussed. New mechanisms of atomic transport are pointed out, explaining the enhancement of the SiO<sub>2</sub> precipitation process and then the extension to the silicon bulk of the SiO<sub>2</sub> precipitation region.

#### 2. EXPERIMENTAL CONDITIONS

A polysilicon layer with thickness of about 1500 nm was grown by low pressure chemical vapour deposition (LPCVD) at the pressure of 0.2 torr and the temperature of 620°C on p-type, (111)-oriented, Cz silicon wasers with a resistivity of (8–12) Ωcm. The polysilicon layer was phosphorus doped by diffusion from a liquid POCl<sub>3</sub> source at the temperature of 1030°C for 15 min. After the phosphorus prediffusion, the polysilicon heat treatment was continued by a subsequent annealing for 20 min at the same temperature of 1030°C to drive-in the dopant by a diffusion process into the silicon.

Three types of samples were investigated: specimen 1 with the polysilicon layer as deposited, specimen 2 after phosphorus prediffusion from POCl<sub>3</sub> source at 1030°C for 15 min and specimen 3 after drive-in diffusion by annealing at 1030°C for 20 min. After each step of annealing there were determined the junction depths in silicon to observe the doping conditions of the thick-polysilicon system. The junction depth of the specimen 2 and 3 were determined by cross-sectional electron-beam induced current (EBIC) measurements with a JSM 840 scanning electron microscope.

In order to correlate the structural modifications of the polysilicon layer with the phosphorus diffusion conditions, the specimens of type 1, 2 and 3 were also examined by a JEM 2000 FX transmission electron microscope (TEM) to delineate the structural morphology of the polysilicon and polysilicon/silicon interface.

#### 3 RESULTS

The main results obtained from TEM determinations are shown in Figs. 1-3. Fig. 1-bows a TEM cross-sectional micrograph of the thick polysolicon/solicon.

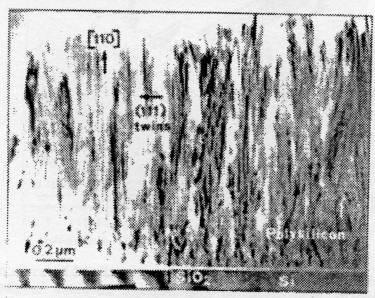


Fig. 1 - XTEM cross-sectional manageraph of the polysidacon /SiO<sub>2</sub>/sidacon system after the LPCVD process at 620°C.

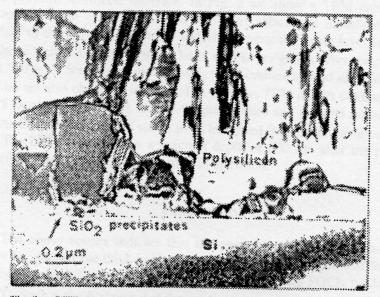


Fig. 2 - XTEM cross-sectional micrograph of the polysitions /SiO,/ willcon system after the phosphorus prediffusion at MMPC for 13 min.

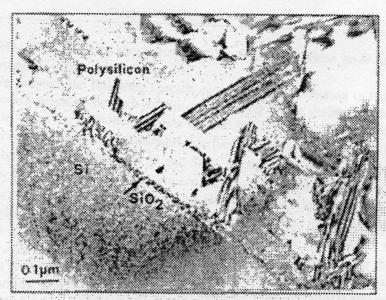


Fig. 3 - XTFM cross-sectional interrugraph of the polysition /SiO Advices system after the phosphorus drive-in diffusion at 1030°C for 20 min.

system after the LPCVD process. A columnar structure of [110] oriented grains are observed within the deposited thick polysition layer, characterising its initial morphology. A continuous SiO<sub>2</sub> film of thickness of about 16 nm is present at the polysilicon/silicon interface. In Fig. 2 we show a TEM cross-sectional micrograph of the type 2 specimen exhibiting substantial modifications in the structure of the polysilicon/silicon and SiO<sub>2</sub> layers; the oxide layer is broken up and the polysilicon is regrown to a layer with larger grains (average grain size of about 300 nm). Some local regions with precipitations instead of a SiO<sub>2</sub> continuous film are also observed in Fig. 2. Fig. 3 shows TEM results obtained from the type 3 specimen exhibiting local SiO<sub>2</sub> precipitates and isolated islands of SiO<sub>2</sub>. The thickness of this region seems to be increased from 16 nm to about 20 nm.

The EBIC results show a junction depth of 6(x) nm after the phosphorus prediffusion and of 8(x) nm after drive-in diffusion.

#### 4 DISCUSSION

The above results indicate that the phosphorus doping process is sufficiently fast so that the thick polysilicon layer to be fully doped and therefore to become a local diffusion source for silicon after 15 min POCI, prediffusion at 1030°C. Such a result shows that the moderate temperature (1030°C) short-time duration (15 min) phosphorus diffusion annealing is sufficient to break-up the thick  $SiO_2$  film, allowing the regrowth of the thick polysilicon layer and then the phosphorus diffusion in silicon.

We propose the following specific mechanisms contributing to the structural modifications of the thick polysilicon layers on silicon during the phosphorus diffusion in the polysilicon/silicon system, leading to the SiO2 breaking-up and precipitation near the SiO<sub>2</sub>/silicon interface; (i) due to the different thermal expansion coefficients between Si and SiO2, strong stress is developed within the SiO2 film which is confined between the thick-polysilicon layer and the Si substrate; the high strain fields could be relaxed by SiO2 mechanical damage like cracks in local weak regions; such weak regions could be created by local conversion of SiO2 into phosphosilicate glass (PSG) as a consequence of the incorporation of high phosphorus concentration (>1%) by fast diffusion through the grain boundaries: (ii) the silicon self-interstitial injection into the polysilicon layer, coming from two sources: (1) the oxidation of the polysilicon surface and the conversion into the phosphosilicate glass, in the ambient of POCl<sub>3</sub>, O<sub>2</sub> and N<sub>2</sub> gases during the phosphorus prediffusion: (2) the phosphorus diffusion itself, contributing to an efficient and short-time regrowth into a larger grain size structure, induces a rapid SiO2 degradation as we will present below.

As the first type of mechanism is concerned, we have to take into account that the excess phosphorus atoms diffusing from the surface along the columnar grain boundaries should be effective especially during the first stages of the prediffusion, when the density of the fast diffusion paths is sufficiently high and short. Therefore, according to the proposed mechanism (i), we have to expect an efficient process of breaking-up of the SiO<sub>2</sub> layer during the phosphorus prediffusion (Fig. 2).

The SiO<sub>2</sub> precipitation observed from Fig. 2 and Fig. 3 can be explained by the silicon self-interstitial participation at the local atomic transport. Due to the excess of silicon self-interstitial atoms coming from the phosphorus diffusion and from PSG growth process, a simple reaction including SiO<sub>2</sub> decomposition occurs as:

$$Si + SiO_2 = 2SiO. (1)$$

SiO is a mobile species [5] with a diffusion coefficient in the SiO<sub>2</sub> layer of the order of 10<sup>-15</sup> cm<sup>2</sup>/s at 1030°C. This species can diffuse away from the polysilicon/SiO<sub>2</sub> interface and precipitate by the reverse reaction, or can be lost by diffusion along the grain boundaries (Fig. 3). Such a process determines the degradation of the SiO<sub>2</sub> film continuity, even for such a thick SiO<sub>2</sub> film. SiO migration through the initial SiO<sub>2</sub> film is a limiting process during the first stages of the heat treatment (Fig. 2), but during the phosphorus diffusion, SiO<sub>2</sub> layer becomes a more and more open structure for the diffusion, so that the SiO

flux penetrates easier in the silicon substrate, carrying out O atoms to the silicon bulk. Near the SiO<sub>2</sub>/silicon interface, where the local equilibrium of reaction (1) is modified, the reverse reaction occurs.

$$2SiO = SiO_2 + 2Si,$$
 (2)

where Si is the silicon self-interstitial atom.

Reaction (2) requires two molecules of the same type and therefore is likely to occur just near the polysilicon/SiO<sub>2</sub> interface, where the SiO concentration ( $C_{SiO}$ ) is sufficiently high. Reaction (2) then describes a short-range SiO<sub>2</sub> precipitation effect. Such a reaction determines the SiO<sub>2</sub> degradation on the basis of the consumption of the oxygen atom content from the SiO<sub>2</sub> interfacial layer. In the bulk of the FZ silicon wafers no oxygen atoms are present and therefore only reaction (1) is operative for SiO<sub>2</sub> precipitation.

As the phenomena described above already suggest, the existing oxygen atoms in Cz silicon wafers may influence the structural properties of  $SiO_2$  films located between the polysilicon and silicon during the phosphorus diffusion. As it was mentioned before, a flux of SiO molecules coming from the polysilicon/SiO<sub>2</sub> interface transports the oxygen atoms to the silicon substrate. In the Cz silicon wafer a flux of oxygen atoms dissolved during the growth process and coming from the silicon bulk (where the oxygen concentration  $C_0$  is equal to the initial value  $C_i$  of the oxygen concentration) has to be taken into account [6]. We suppose for the simplicity that the oxygen concentration is lower than the critical value for the bulk precipitation (Fig. 4). An additional reaction way occurs

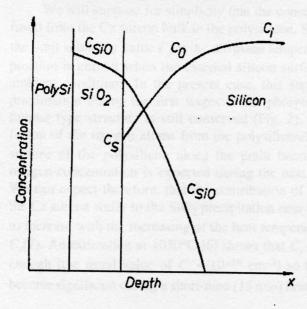


Fig. 4. – Schematic representation of the oxygen concentration and of the concentration of the SiO mobile species in Cz silicon near polysilicon/SiO, interface.

under these conditions between SiO molecules escaped from reaction (2) and the oxygen atoms coming from the Cz silicon bulk:

$$SiO + O = SiO_2. (3)$$

This reaction does not contribute directly to the degradation of the initial continuous  $SiO_2$  film, but enhances the  $SiO_2$  precipitation process near the polysilicon/SiO<sub>2</sub> interface. Moreover, reaction (3) produces a long-range precipitation because of the oxygen excess in the region near the polysilicon/SiO<sub>2</sub> interface, which is of the order of the SiO diffusion length. Relation (3) describes then a long-range precipitation effect.

An additional enhancement of the SiO<sub>2</sub> precipitation in Cz silicon wafers is induced by the reaction between the bulk oxygen atoms and the silicon self-interstitial atoms coming from: (i) phosphorus diffusion in silicon substrate; (ii) phosphorus diffusion in polysilicon grains; (iii) polysilicon oxidation during the prediffusion step; (iv) conversion of SiO into SiO<sub>2</sub> (reaction (2)), under the form:

$$Si + O = SiO.$$
 (4)

This reaction does not contribute to the SiO<sub>2</sub> film degradation, because no oxygen atoms from the interfacial SiO<sub>2</sub> film are used. By means of reaction (3) and/or (2), SiO<sub>2</sub> precipitates are generated as follows: during reaction (3) silicon self-interstitial and oxygen atoms are consumed, while during reaction (2), the silicon self-interstitials become free atoms and may participate again in to the atomic transport in silicon.

We will suppose for simplicity that the concentration of oxygen atoms diffused from the Cz silicon bulk to the polysilicon. SiO2 interface is the same with the limit solubility value  $C_s$  at the diffusion temperature [6] (Fig. 4). Such a supposition is correct when the external silicon surface is exposed directly to the ambient conditions. In the present case, this supposition may be a good approximation during the first stages of the phosphorus diffusion, when the columnar type structure is still conserved (Fig. 2), because of the rapid out-diffusion of the oxygen atoms from the polysilicon/SiO<sub>2</sub> interface to the external surface of the polysilicon along the grain boundaries. An increasing of the oxygen concentration is expected during the next stages of the heat treatment. We can expect therefore, that the contribution of the bulk dissolved oxygen in the Cz silicon wafer to the SiO<sub>2</sub> precipitation near the polysilicon/SiO<sub>2</sub> interface to increase with the increasing of the heat temperature, depending on the values  $C_s(T)$ . An estimation at 1030°C [6] shows that  $C_s \sim 2 \times 10^{17}$  cm<sup>-3</sup>, which is high enough (the usual value of  $C_i = 10^{-18}$  cm<sup>-3</sup>) so that the precipitation process become significant during a short-time (15 min) heat treatment at 1030°C (Fig. 2).

During the phosphorus diffusion, the SiO2 precipitation process occurs especially near the SiO<sub>2</sub>/silicon interface, in the short-range region. A continuous atomic transport from the SiO2 film and from the Cz silicon bulk contributes to the SiO<sub>2</sub> precipitation as follows: (i) the consumption of oxygen atoms by means of reactions (1) and (2) which determine the degradation of the initially continuous SiO2 film and the SiO2 "regrowth" in a silicon region near the SiO2 interfacial layer; (ii) the oxygen atoms coming from the Cz silicon bulk, determine by means of reactions (3) and (4) both the enhancement and the spatial extension (long range effect) of the SiO<sub>2</sub> precipitation process. The two kinds of atomic transport are related by the oxygen atoms coming from the SiO2 film (reaction(1)) and from the Cz silicon bulk, reaction (2) and (3). The new SiO<sub>2</sub> formed regions will interact again with silicon self-interstitial atoms generated during the phosphorus diffusion and an additional cycle as described by reactions (1)-(4), will contribute to a further extension of the SiO<sub>2</sub>-precipitation region. It can be therefore explained in such a way the conversion of the SiO<sub>2</sub>. interfacial film from 16 nm thickness (Fig. 1) to a 20 nm thickness of a SiO<sub>2</sub> precipitated region after 35 min phosphorus diffusion at 1030°C (Fig. 3).

#### 5. CONCLUSIONS

The short-time (15 min) moderate temperature (1030°C) heat treatment of thick (1500 nm) LPCVD polysilicon/SiO<sub>2</sub> on silicon during the phosphorus diffusion from a liquid POCl<sub>3</sub> source is sufficient to fully dope the polysilicon layer and to convert it in a local diffusion source. The XTEM determinations after the LPCVD deposition, phosphorus prediffusion and phosphorus diffusion allow to show that:

- (1) a degradation of the initial thick SiO<sub>2</sub> occurs during the phosphorus diffusion at 1030°C for 15 min: the SiO<sub>2</sub> film is broken-up and SiO<sub>2</sub> precipitates are formed:
- (2) a further degradation of the  $SiO_2$  layer occurs during a phosphorus diffusion step at  $1030^{\circ}$ C for 20 min, consisting in the extension of the  $SiO_2$  precipitation region from 16 nm after the LPCVD to 20 nm thickness after the diffusion;
- (3) the SiO<sub>2</sub> film degradation was explained by the oxygen consumption from the SiO<sub>2</sub> film by means of reaction with the self-interstitial atoms mainly generated by the phosphorus diffusion and by the polysilicon oxidation;
- (4) the enhancement of the precipitation process near the SiO<sub>2</sub>/silicon interface in the Cz silicon wasers and the extension of the SiO<sub>2</sub> precipitation

region was explained by the atomic transport of the oxygen atoms dissolved in the bulk of the Cz silicon wafers, diffusing to the SiO<sub>2</sub>/silicon interface and interacting with Si and SiO species near the SiO<sub>2</sub>/silicon interface.

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